(8737) C. F. H. asks: Can you give INDEX OF me any information as to the mixture used in binding coal screenings together that are made into briquettes? A. The best material for binding coal fines into briquettes, and the one most largely usea, is pitch. Asphalt has had a limited use. Starch paste, residues from starch manufacture, dextrine, molasses, etc., have been used from time to time experimentally, but are not practicable. Various mineral substances, such as clays, lime, water-glass, etc., have also been proposed, but naturally have the drawback of adding just so much ash. Occasionally, oxidizing materials, such as niter, are added, when it is desired to produce a very quickly burning briquette for the rapid generation of high temperatures.

(8738) W. J. C. wishes to know how to remove indelible ink marking from clothing. A. Indelible inks are of such variable character that it is quite impossible to reply. Many of these inks have nitrate of silver as a basis; in this case. a solution of hyposulphite of soda might help. Some other inks might possibly be bleached out with javelle water and weak muriatic acid; this can be used only on white goods, as most dyes would be destroyed. Possibly also a solution of sulphurous acid might be of service.

(8739) S. R. asks for a good receipt for making a reliable fire extinguisher in powder form, one that is easy to prepare. A. For a cheap, dry powder fire extinguisher, bicarbonate of soda will serve; it may advantageously be mixed with 5 per cent to 10 per cent in some powdered mineral, as flint, tripoli, chalk, etc., to prevent caking in damp air. A mixture of dry bicarbonate of soda with dry sal-ammoniac, and kept in a dry place, will do better, as it would yield both carbonic acid and ammonia. In a confined space fire extinguishers of a type similar to ${f gunpowder\ have\ proved\ effective}$; the object being to fill the room with carbon dioxide, sulphur dioxide and nitrogen gases and thus choke the fire. A good formula for this type of extinguisher is niter, 60 parts; sulphur, 36 parts, charcoal, 4 parts.

(8740) W. R. asks what the different gases are which, if introduced into an inclosed arc lamp will turn the color red, green, vellow, blue, etc. A. Colored electric lights are ordinarily produced by coating the globe with an aniline dye, made in alcoholic solution, and mixed with a little varnish. We do not know any gas which could withstand the heat of the arc for any time and which could color the arc. Some color can be imparted to the arc by soaking the carbons in solutions of sodium chloride, strontium chloride, or lithium chloride, and drying them thoroughly before using. The light of the arc itself is so intense that it is very difficult to overcome it with any other colored light.

(8741) H. M. asks: Can you give me information as to what a transformer is and what it is used for? I have been informed that it is much on the scale of an induction coil. If so, can you give me some scale by which to transform a 110-volt current into amperes? A. A transformer changes an alternating current from one voltage to another and from one current strength to another. It cannot change volts into amperes. In that respect they resemble induction coils. An induction coil is a particular sort of transformer, provided with a condenser, interrupter, etc. It is used almost entirely for raising the voltage. You will find a good chapter upon Transformers in Houston and Kennelly's "Alternating Currents." 2. Also please tell me how many volts it will take to each ampere, and a scale of how it should be wound, what size wire to use, and if the fine wire should be used outside or in? A. It is impossible to change amperes into volts. And as to the winding, each one is wound for the work it is to do There is no general winding.

(8742) G. W. L. asks: 1. What is the most economical method of generating carbonic acid gas-not necessarily pure-in large quantities? A. The commercial sources of carbonic acid, on a manufacturing scale, are as follows. 1. By the burning of limestone. 2 By the action of acids in limestone (calcium carbonate), magnesite (magnesium carbonate) or dolomite (calcium magnesium carbonate) The acid used is sulphuric. This method is used by the manufacturers of bottled effervescing waters. 3. By collecting the carbonic acid gas generated in the fermentation vats of large breweries. This source is largely used in Germany. In addition, the gas coming from many of the natural springs is collected. This practice is also largely used in Germany. 2. Are there any known chemicals, or other substances, that will decompose water, aside from the alkaline metals? A. Besides the alkaline metals, water is decomposed by many of the hydrides and carbides of the different metals. Thus calcium carbides decompose water with the formation of lime and acetylene. Also, vapor of water passed through red-hot tubes of different metals is decomposed into its con stituents. Vapor of water passed through red-hot coal is decomposed, with formation of carbon monoxide and dioxide, hydrogen, marsh gas (CH,) and other hydrocarbons; this is the basis of the industrial manufacture of water gas, which has displaced coal gas in most cities.

INVENTIONS

For which Letters Patent of the United States were Issued for the Week Ending

October 28, 1902,

AND EACH BEARING THAT DATE. [See note at end of list about copies of these patents.]

	Acid, making formylmethylanthranilic, F. Von Bolzano	712,246 12,047	
	Air brake, J. B. Briggs, Jr	712,249	
l	Acid, making formylmethylanthranilic, F. Vom Bolzano Air brake, J. E. Norman, reissue Air brake, J. B. Briggs, Jr. Adjustable switch, E. W. Harden et al T. Adjustable switch, E. W. Harden et al Agricultural rake, A. R. Black Alarm. See Automobile alarm. Alloy of silver, W. H. Walker Alum, obtaining, H. H. Wing Amusement device, A. T. Prescott Animal trap, J. E. Cox Apron, W. T. Jones Armature for dynamo electric machines, J. Burke Atomizer, A. H. Tatum Atomizer, A. H. Tatum Automobile alarm, pneumatic, G. E. Cordeau C. L. Williams, F. Lau Bag or satchel frame, F. Lau Bag or satchel frame, F. Lau Bale tie, J. W. Griggs. Baling press, P. C. Southwick Barrel or cask, H. P. Lemper. Battety plate separator, storage, R. N. Chamberlain Bean cutter and puller, C. H. Knapp. Bearing, side, C. H. Williams, Jr. Bed, folding, C. C. Taylor Bet, distributor or conveyer, J. F. Tinsley W. W. Beberts, device for applying, W. M. Beberts, device for applying, M. S. Beberts, M. S. Bebert	712,092 712223 711,933	
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ı	ley Belt dressing to belts, device for applying, W. K. Richart Belt holder, G. D. Schrum Bier, knockdown, W. M. Rankin Billet heating furnace, continuous, H. B. A. Keiser Binder temporary E. L. Krag	712,152	
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	Burners. See Hydrocarbon burner. Busy test system, D. S. Hulfish. Button, J. M. Marks. Button, W. C. Vogel. Cable signal, traveling, M. Norden. Calcium sulfate and by products, obtaining, H. H. Wing. Calculating machine, J. T. Howieson. Camera, divided bed photographic, Robertson & Hutchings Camera, swing-back photographic, C. E.	712,080 712,426	l
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ı	Can dle stick or holder, A. Tovey	712,302 712,396 712,216 712,327	
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and making same, brown sulphur, M.

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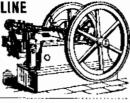
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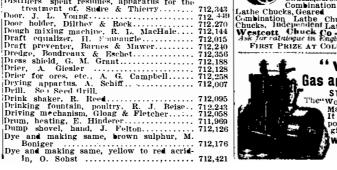
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1		711,953
٠	Dynamo, exciting, H. W. Buck712,041, Electric accumulator, F. Loppe et al	712,451 712,316
١	Electric accumulators, forming, W. Morison Electric circuits, automatic switch for fuse	711,997
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Į		712,052
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į	Envelop, M. Stormfels Envelop, mailing, M. J. Daly	712,465 711,945
١	Electrical distribution system, C. P. Steinmetz	711,992
į	Exercising device, B. A. McFadden Explosive engine, C. E. Inglis	712,000 712,067
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į	Feed water regulator, Flinn & Aborn. Feeder, boiler, I. S. Davis. Fence, E. D. Carr. Ferrule cover, trap, O. G. Hitchcock. Fertilizer and the product resulting therefrom, converting salt marsh material, etc., into a, J. J. Crooke. Fibrous materials, machine for balling or winding slivers of, W. B. Lee. File, card index, C. W. Scarritt, reissue. File, loose leaf, L. G. Schult. File, loose leaf, L. G. Schult. Firearm magazine, A. D. Marble.	112,510
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ļ	File, letter, F. Tramblay File, loose leaf, L. G. Schult	12,048 $712,217$ $712,008$
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į	Meat, Making a preparation for preserving, Markovits & Guthmann	711,990
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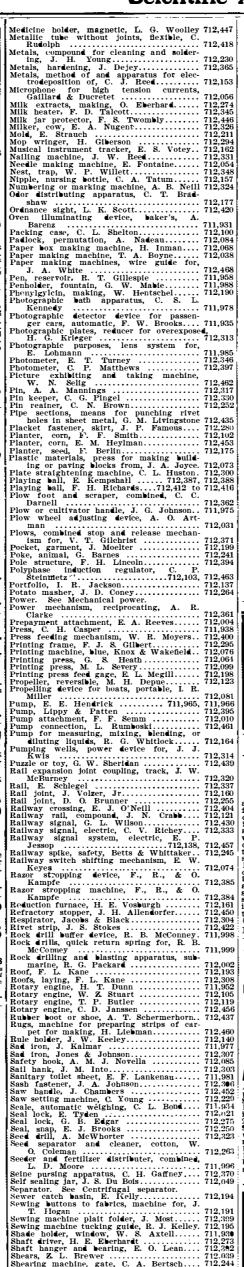
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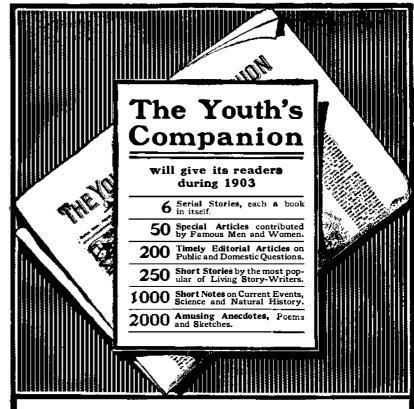
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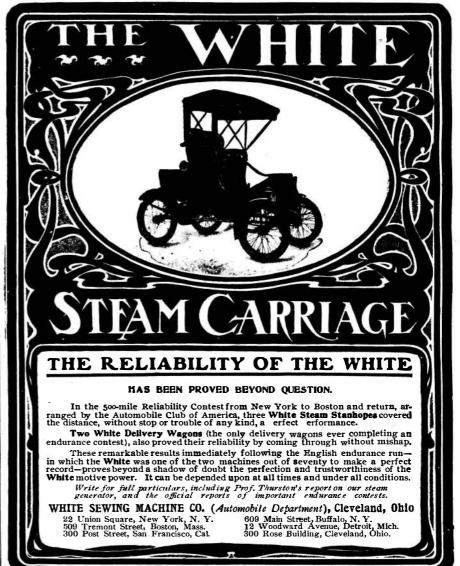


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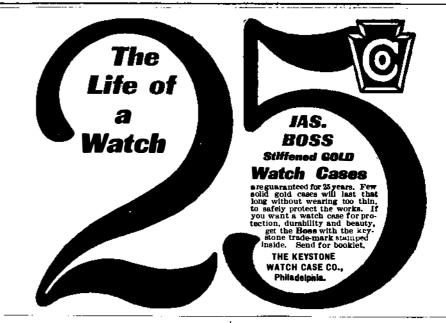
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