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Mineral Lands Prospected, Artesian Wells Bored, by Pa. Diamond Drill Co. Box 423, Pottsville, Pa. See p. 46.

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on the recovery of tin scrap to profit, as well as an abundance of scrap.

(2) W. P. M.-Galvanized iron chain pumps do not affect the water perceptibly for drinking or cooking purposes. They are largely used in wells and cisterns.

(3) J. I. V. D. asks the formula and apparatus necessary for etching pocket cutlery. A. Write with a chloride of gold solution. 2. A brown ink for writing on polished steel. A. We would suggest the following as giving you a colored ink: sulphate of copper 1 ounce, sal ammoniac 1/2 ounce, pulverize separately, adding a little vermilion to color it, and mix with 11/2 ounces vinegar. Rub the steel with soft soap, and write with a clean hard pen without a slit, dipping in the mixture. To produce a bronze color, it would be necessary to first use a black ink, and subsequently coat with bronze powder.

(4) B. P. T. asks: 1. What material is used to bleach skeletons? A. See "Preparation of Skeletons for Museum Purposes "in Scientific Ameri-CAN SUPPLEMENT, No. 106. 2. What are the "green soaps" made of which are used in skin diseases, and who manufacture them? A. Green soaps are prepared from potassa and fixed oils. They are officinal, and therefore can be procured from any druggist. 3. What is the mucilage used on the back of postage stamps 1 part.

(5) C. P. asks what is the commercial hydrocarbon gas black, and how is it made? A. The preparation of gas black is probably similar to that of lamp black, and is therefore produced by burning ordi nary illuminating gas in a supply of air which will be in-sufficient to completely oxidize it. The black fumes arising from this combustion deposit the carbon on cloths or, in some instances, on metallic plates suspended in order to receive it. Gas carbon, which is possibly what you refer to, may be obtained by passing olefiant gas through a red hot porcelain tube.

(6) H. R. H. asks (1) for a practical recipe for the generation of oxygen gas, for the purpose of inhalation. A. See "How to Make Oxygen," in Sci-ENTIFIC AMERICAN SUPPLEMENT, No. 313. 2. What effect would the inhalation of sulphur fumes have on the system? A. The inhalation of sulphur has been recommended for the curing of contagious throat diseases, such as diphtheria, etc. The fumes are suffocating, and care must be taken in their inhalation.

(7) E. H. F.—Tonics or washes to make the hair grow can always be employed with greater or less success so long as there is any vitality left in the hair roots. If, however, these are destroyed or entirely dead, there is no possibility of producing a new growth of hair. The following is a well known tonic: Scald black tea 2 ounces with 1 gallon of boiling water, strain, and add 3 ounces glycerine, tincture cantharides 1/2 ounce, bay rum 1 quart. Mix well, and perfume.

(8) J. W. E. A. asks whether an analysis has ever been made of any of the petrifying lakes or wells in Europe or elsewhere, with a view to iscover the petrifying properties of the water, and, if so, what was the result? A. Calcium carbonate. which is insolu ble in pure water, but soluble in water containing car bonic acid, is frequently found in springs of carbonated waters which pass through limestone. Objects placed in such waters are coated with a deposition of the carbon ate of lime. Travertine is the name given to such formations. Silica is likewise dissolved by certain mineral waters, and the deposits obtained by the evaporation of such waters are known as silicious sinter. Works on geology fully describe the process. 2. What chemical or combination of chemicals, apart from those ordinarily nsed in refrigeration, will freeze liquids? A. For this information, see table given in answer to query 4, in SCIENTIFIC AMERICAN for June 21, 1884.

(9) T. F. W. asks what to put with pure white paraffine wax to make it pliable long enough when dipped in water to make imitation of roses. A. Use pure beeswax (white), and mix with paraffin.

(10) W. A. W. writes: Suppose two boilers 200 feet apart are connected by a 2 inch pipe, fire beneath one only, and gauge shows 150 pounds pressure. Would a gauge on the other boiler indicate same or lcss pressure, and if less, about what per cent? And would size of pipe or distance of boiler cut any figure? A witness in court testified pressure would be materially less in the distant boiler, owing to condensation, but I would prefer to have you say whether he is correct before I can believe so. A. The gauge on the farther boiler would indicate at first considerably less pressure, until the boiler and pipe had become heated by the condensing steam, when the gauges would indicate less difference, but the pressures would never be quite equal. A great deal depends upon the amount of surface exposed in pipe and boiler, and the outside temperature, in determining the per cent difference Boiler furnace, J. Mailer.....

double bass, something with lots of "hold fast" in it. Buckle, suspender, J. R. Pollock..... The objection^{*} to most which is kept for sale in music houses is its tendency to prevent vibration rather than increase it. A. If the resin is too sticky, as we infer from your communication, the best thing to do is to reboil it, when it will be found more satisfactory. If too hard, keep it in a warm room for several days

(14) L. L. asks for formulas for making ordinary blue prints. A. See SCIENTIFIC AMERICAN of October 31, 1885, Photographic Notes, page 276.

(15) S. desires a recipe for cleaning micas in stoves. A. It is not possible to perfectly restore the micas after they have once been burned. Rubbing with a little alcohol and water may improve them slightly.

(16) J. G. H. asks the best means of precipitating lead in solution in strong water of ammonia without injuring the water of ammonia. A. Metallic zinc has the property of reducing lead to its metallic state when in solution. This method does not always give satisfaction. For the complete separation of the lead, it is best to treat the solution with the gas of hydrogen sulphide. See Fresenius' Qualitative Analysis.

(17) Machinist asks for a recipe to renew an old rubber coat or gossamer. A. Brush over with a solution of 1/2 ounce of pure rubber dissolved in made of? A. Gum dextrine 2 parts, acetic acid 1 part, 1 pint of carbon disulphide. By proper treatment and water 5 parts. Dissolve in a water bath, and add alcohol a series of coats a deposition of rubber on the fabric vill be obtained. See "Rubber Waterproof Goods," in SCIENTIFIC AMERICAN SUPPLEMENT, NO 251.

(18) W. W. K.-Alcohol contains 91 per cent ethyl alcohol by weight, and has a specific gravity of 0.820 at 60° Fah. Proof spirit contains 49.24 per cent alcohol and 50.76 per cent water, and it has a gravity of 0'919 at 60° Fah. In other words, it is 50 per cent alcohol. Salep is the name given to dried timbers of numerous species of the genus Orchis, and in India of the genus Eulophia. Leading druggists can obtain it for you.

MINERALS, ETC.-Specimens have been received from the following correspondents, and examined with the results stated.

A. C. V.-The specimen is a white clay or kaolinite. It is valuable in the manufacture of pottery. It is found invarious sections of the country. We would advise you to send it to some maker of pottery in your vicinity, in order that its burning qualities may be tested .-- J. M.-The reddish specimen is a variety of clay, and from a superficial examination we think it has the properties of a fire clay. It would be necessary to test its burning qualities in order to positively determine this fact. The powder sent is likewise a clay containing iron and probablymanganese. It does not seem to be of any value C. H. C.- The specimen sent is a bit of manufactured wire, and is probably a piece of the drill which in some way has been broken off at the end .-- W. A .-- The powder is simply a clay ighly colored with metallic oxides, probably iron.

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AND EACH BEARING THAT DATE.

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give date of	paper and page or number of question.			Fintering apparatus, W. Olunam
Inquiries no	ot answered in reasonable time should	The witness was correct.	Boiler furnace, steam. S. T. Owens 330,147	Firearm. breech-loading, S. A. Sullenberger 330,354
be repeated;	correspondents will bear in mind that	(11) G. F. F. asks how to mix South	Boiler, connecting box for water tube, W. Kent 330,128	Fire escape, J. Flietner 330,103
some answe	ers require not a little research, and,		Boilers, making connecting boxes for, W. Kent 330,129	Flag, J. M. Ebersole 330,099
	ndeavor to reply to all, either by letter partment, each must take his turn.		Bolt føstener, E. G. Holden 330,119	Flanging machine, R. Munroe 330,141
	prima ion requests on matters of	to analyze 14 to 151% per cent of available phosphoric	Bookbinding machine, Durkee & Campbell 330,303	Flax, tank for curing, I. T. Quinn 329,949
	her than general interest, and requests		Boot or shoe, M. Walker	
for Prom	at Answers by Letter, should be	parts of the rock and 1,000 tons of sulphuric acid. 2.	Boot or shoe heels, cutter head for trimming, B.	Folding chair, E. L. Gaylord
accompanied	with remittance of \$1 to \$5, according	What grade of sulphuric acid? A. Sulphuric acid		Folding chair, Hall & Tripp
to the subject	ct, as we cannot be expected to perform	of 50° Baume. 3. Is there any artificial way of drying		
	without remuneration.			Forging machine, radial, J. C. Richardson
to may be h	merican Supplements referred ad at the office. Price 10 cents each.		13 000 1/0	
	nt for examination should be distinctly	South Carolina rock phos. as high in phosphoric acid	Details and can weather II Reather 990 PE	
marked or la		when just mixed as it will be after lying a month, or	Bottle stopper, A. Luedemann	
		does age improve it? A. The phosphoric acid reverts	Bottling machine, H. C. Walter	Glass flattening furnace. Tinner's portable
(1) D. D	-The recovery of tin from	if allowed to rest for any length of time.	Box. See Blacking box.	furnace.
	ticed by several chemical manufactories			Furnace, E. Boileau
	New York, but not in the metallic state.	(12) L. H. M.—For information on lu-		Furnace for melting glass, etc., Pearson & Kit-
	-	bricants, see Scientific American Supplement, No.	Brace. See Rail brace.	
•	burned and thoroughly oxidized by fire	316 For avie grease, Dissolve 14 nound common soda	Brake. See Marine brake. Velocipede brake.	son
with a free circ	culation of air. Then treat the oxides	in 1 gallon water, add 3 pounds tallow and 6 pounds	Brewing beer, C. Zimmer	
with sulphurica	cid, which unites with the iron, forming	•	brick machine, b. Daig	hearth, Murisier & Gilchrist 329,937
sulphate of iron.	which may be decanted and crystallized	palm oil (or 10 pounds palm only). Heat them to-		Gauge. See Railway track gauge.
• •	into rouge, or the red oxide of iron,	getber to 200° or 210° Fah., mix, and keep the mixture		Gauge and center square, combined, J. C. Eckert. 329,892
	uing tin sediment may be further oxi-	constantly stirred till the composition is cooled down		Gas apparatus, R. H. Smith 330,165
		to 60° or 70°.	Hatch 830,111	
	into polishing powder, or, as by some		Brush, clothes, J. Stehlin 330.270	Isbell
chemists, made	into sulphomuriate of tin, as used by	(15) D. D. L. desires a formula for mak-	Brush handle, C. Donaghy 330,005	Gas distribution. R. H. Smith 330,267
dyers. It requi	ires some chemical knowledge to carry	ing a good applicant (resin or rosin) for the bow of a	Buckle, M. L. Hall 330,232	Gas engine, W. A. Graham