- (27) C. Y. M. asks whether a long coupled of batteries, electro-magnets, and short stretches of out- | MENT, No. 252, and would thank you to explain the the best and cheapest battery to use to work on a line wagon pulls easier or harder than a short coupled, or whether it is the same; and if one pulls easier than the other, why? A. The long coupled wagon will draw the pose; but one provided with a horizontal needle would shows both of same diameter; which is right? A. The easier, because the horses will have a greater leverage over a twisting strain, caused by an obstruction on one side of the road.
- (28) W. G. R. asks how to prepare thin ever. paper that, by writing on it with a pointed stick while lying on white cloth, the writing will be transferred to the cloth. Would like to have it stand washing. A. We do not know of an indelible transfer paper.
- (29) H. G. M. asks the formula of a composition used in soldering to make solder stick. A. Dissolve zinc in muriatic acid until effervescence ceases; dilute it with a quantity of water equal to that
- (31) A. C. H. asks: What are the ingredients of gold solutions that deposit a red color on metal to be plated? I use the Smee battery; has that any effect on the coloring or gilding in this case, or what battery is best to use? A. A Smee's battery is as good as any for the purpose. For information on electrometallurgy, consult SUPPLEMENT, No. 310.
- (32) C. L. W. writes: Our house is alive with fleas from apet cat. We have tried Persian insect powder without much effect; can you tell me of anything better for the purpose? A. The oil of pennyroyal will drive your fleas away. Beds made of pennyroyal for dogs to lie on will also drive away fleas.
- (33) G. W. S. asks a receipt of some liquid or oil to place in gravity batteries that will keep them from evaporating and using up so quick? A. Any nondrying oil will do; try common lard oil.
- (34) G. S. T. writes: I have an eighty barrel water tank on barn; the bottom of tank is 24 feet above the ground. A windmill supplies the tank antomatically. My carpenter's shop floor is 26 feet below low water mark in tank and about 6 rods in a lateral direction; with a two inch pipe connecting tank to a suitable turbine at the shop, what amount of power ought to be developed, and in what time 80 barrels be run out, when turbine is at full work? A. The discharge of the 80 barrels of water will develop ¾ horse power for an hour, or 1/4 horse power for 2 hours. You will not realize more than 60 to 70per cent of this power through a motor.
- (35) Engineer writes: I have some celluloid billiard balls that have become faded. I wish to recolor them. What shall I use? A. Use the following colors in the proportion of about 10 grains to the gallon of water: Black, use nigrosine; for violet, methyl violet; blue, soluble blue; red, anil ne red or magenta; green, malachite green. Dissolve these aniline colors in hot water. The balls are immersed in the hot water, and then al. lowed to cool in the coloring solutions until sufficiently dyed. Perhaps three to four hours.
- (36) J. P. B. writes: Please give me a recipe for making the following inks, aniline preferred: until the coloring disappears. This reaction will show green, blue, burnt sienna, yellow and purple. A. Green: Dissolve 1 part of iodine green in 100 to 110 green; if it is desired to give it a yellowish green tint, for freeing the first equivalent? A. The use of hydrate alittle picric acid is to be added. Blue: Dissolve 1 of lime would not be as effective as the carbonate. part of the soluble bleu de nuit (bleu de Paris) in 200 to The tartaric acid decomposes the lime carbonate, and 250 parts of hot water. Yellow: Dissolve 1 part of picric acid in 120 to 140 parts of water. Purple: Dis- hydrate was used. 4. Can the sulphur be freed from solve methyl violet in sufficient water. For brown ink the potassium sulphate? If so, how? A. Barium chloride select a suitable aniline color that is soluble in water, will precipitate the sulphur from potassium sulphate, and add a small quantity of alcohol and a little gly- forming barium sulphate and potassium chloride, 5 cerine (1 to 4 per cent). The addition of these two ingredients is desirable inany case,
- (37) E. S. writes: We have lately had our cistern cleaned. The water seemed perfectly clear and pure, without taste or odor; but as it had gotten quite low, and as the cistern had not been cleaned for several years, we thought it best to have it emptied and thoroughly cleaned. Since then, the water has had a very bad taste and smell, which we can notice even after it has been through the filter. Can you tell me why this is so? A. In cleaning your cistern you may have exposed the clean cement to the action of the water, which may have given it the taste of lime or alumina. As you do not tell us how it tastes or smells, we cannot well solve the mystery. The soot that was found in the cistern was no doubt derived from the smokeof soft coal, which deposits carbonaceous matter with a little pyroligneous acid upon the roofs where bituminous coal is used. This is washed into the cistern and becomes a deodorand distant nt, finally settling as the black soo mentioned If the water tastes and smells of limeor alumina, it is all right. If of decayed animal matter, it
- (38) A. & F. B. ask if there is such coke manufactured in any part of the States as is used for fuel in English locomotives, and where. A. The coke in the United States, and nearly equal to the best English coke. You will find the names of makers in any commercial agency reference book.
- (39) J. W. K.-There is no special examination required for entering the classes at Cooper Union. You must be 15 years of age. Appear in person, with a letter of recommendation if possible from employer. By calling at the office in second story of needed information.
- (40) S. L. W. writes: Will a Breguet upright galvanometer, such as is sold for \$10, do for experimental work in electric measurement, and also asks course it presently does, a "cancer core "is falsely re will not rust or corrode when frequently used in water how to construct a cheap rheostat to use with the ported, and a reputation for a perfectly inert remedy is and bard enough to bear as much friction (or more) as same. As I am studying electricity out of school hours, I do not wish an expensive instrument for the

- (41) J. S. H. asks: Can you inform me what causes the ivory on piano keys to turn yellow? Some I have recently seen change in a few months from a pure white. Others, many years in use, still retain their original whiteness. Please explain the cause, and suggest a remedy or preventive. A. The yellow color of the piano keys may be due to grease absorbed from the fingers of the player, or it may be that the piano sits in a dark place or is generally closed. Under (30) J. S. writes: I make an axle grease these circumstances ivory is apt to turn yellow. There which I put up in cans. The trade calls for goods put | are also many kinds of ivory, and the inferior qualities np in wooden boxes. I placed some of my grease in do notretain their whiteness without precautions. Good such a receptacle, but cannot prevent the oils penetrating—ivory keys having a liberal exposure to the light ought the wood. How can I treat the boxes in order to avoid—to retain their whiteness for many years. Ivory is that? A. Coat the inside of your boxes liberally with bleached by exposure to sunlight for periods varying from four weeks to six months, or by immersion in turpentine, kept near the surface, and exposure to the sun for three or four days.
 - (42) L. P. A.—We have frequently published articles on microscopy and upon all the modern investigations in this branch of science. It is possible that we may in the future publish elementary articles on the subject.
 - (43) S. E. K. F. writes: I have constructed a small pressure blower as described on page 75 of vol. XXXIX., SCIENTIFIC AMERICAN, and get a fair blast ate of soda in bringing out white lines on blue prints? form absorbs heat from the surrounding air very slowly. througha % inch tube, but I cannot contract it so as to A. The soda bicarbonate turns the picture to a lavendar Making it long and parrow is advantageous, or use two serve as a blowpipe, as I desire. Please give directions sible to so change your blower with any form of pipe as to make a blow pipe with much, if any, additional
 - (44) W. C. M. asks how to refine whale oil soap, so that it will produce a white lather. A. You soap is usually made from the sediment produced in refining the whale oil. 2. What is "English crown olive oil.
 - per cent of bitartrate of potash or cream of tartar can be dissolved in one gallon of boiling water? A. Cream of tartar is soluble in 18 to 14 parts of boiling water. 2. Cream of tartar manufacturers utilize all their waste products, by converting them into tartaric acid. Can I use any other test besides litmus paper, for absolutely determining if sufficient lime carbonate has been used, to perfectly precipitate the first equivalent, and secondly, what is a good test for determining if all the tartaric acid has been freed from the potashin using the sulphate of lime? A. To determine the acidity by other means than with litmus is possible. You can add a little cochineal solution to a portion of the mixture, and then pour in potassium hydroxide (caustic potash) you approximately how much lime carbonate to add. 8. Would hydrate of lime added to very weak solutions of What work in chemistry, published, can I get, that will thoroughly post me in this particular industry? A. There is no literature available on baking powders other than articles found here and there in the SCIENTIFIC AMERICAN and other similar journals. Blyth's book on Foods may contain some information suited to your wants.
- (46) F. G. H. asks if there is any foundation in fact for the prevailing belief that tomatoes cause cancer. And if so, why? Or, rather, how do they operate—by poisoning the blood, or otherwise? A. The belief, which has become quite common, that what is really one of our most valuable vegetables. At else, with proper moderation. Even a good thing may their improvements in 1815. be abused, and a person may become so extremely fond of tomatoes as to consume an excessive quantity, and cause canker sores in the mouth. But it must be understood that there is no resemblance between cancer and canker, except the similarity of spelling. Cancer is a malignant, frightfully dangerous disease; canker is merely a result of disturbance of the stomacb, and is commonly of small importance except from the pain made at Connellsville, Pa., is considered the best made and annoyance it causes. But even for this the tomatoes are not fairly responsible, for though, as stated, it may sometimes follow their very free use, yet with most persons no such result would occur. 2. Have you ever set forth the wonderful power of red clover in curing cancer, a fact indisputable, if taken before death is inevitable? A. The belief in the efficacy of red clover is not very common; it is local, and has no better claim than that concerning tomatoes. Very frequently the institute you may obtain a circular containing all harmless tumors are considered to be cancers; in such cases the various popular remedies—red clover (Trifolium pratense, cancer root (Compholis Americana). etc., are used, and when the tumor disappears, as of established.

- bemoreserviceable, as it would be more sensitive. You internal diameter of the telescope tube should be larger work six or a dozen instruments on the same length cannot readily construct a cheap rheostat without hav- than the clear aperture of the object glass. 2. How of line? I would want the power of the magnets to be ing a standard rheostat with which to compare it. An thick should paper tube be? A. The thickness of the as strong if I had on a dozen as if only one or two, but imperfect instrument of this kind is of no value what- tube is of no consequence. 3. Please give diameter of field and eye lenses. A. The diameters of the field of the line or on the number or strength of the cells of the and eye lenses are unimportant if they are of the correct focal length. 4. Paper reads eye aperture should be 11/4 inches; should it not be 1/4 inch? A. The aperture should be 1/4 inch. 5. What combination of lenses would you recommend for higher and lower power eye For full information on telegraph lines. instruments, pieces than that described? A. It is only necessary to and batteries, consult Prescott's Electricity and Elecpreserve the same relation between the focal lengths tric Telegraph. of the field and eye lenses for higher and lower powers. 6. What gauge of wire is used for gas lighting spark coils? A. Almost any size of wire will answer for this purpose; No. 24 is often used. 7. What should be size of coil? A. The coil maybe 6 inches long, 2 inches in diameter for three or four burners. 8. How many Lecianche cells will be required to light a single burner? A. From two to four cells.
 - (48) W. F. D. asks (1) proportions and ingredients for blue prints. A. Dissolve 40 grains amdissolve 1 drachm potassium ferrocyanide in 1 ounce disferrocyanide produces an intense blue coloration similar to Prussian blue. 3. Chemical action of bicarboncolor, and prevents its fading. A dilute solution of acid and so brings out the white lines.
- (49) W. H. P. asks bow alumina soap is made, and also where I can find a description of the hotair engine. A. There is a soap now manufactured in this country, in which caustic soda is replaced by will be obliged to first refine the whale oil. Whale oil the aluminate of soda. The latter can be prepared either from bauxite or from cryolite. Bauxite is calcined with soda ash, whereby an aluminate of soda is soap?" A. English crown soap is an imported soft formed, and the iron is separated by lixivation, the resoap used by harness makers and the like for rubbing sulting liquors being evaporated until a dry commerand polishing leather. 3. Also soap stock. A. Soap cial aluminate of soda is obtained. Powdered cryolite stock is the residuum from cotton seed oil and from is mixed with six equivalents of lime and boiled with learn to read faceseasily. Would you please inform me water, when an insoluble fluoride of calcium is formed. (45) J. H. G. writes: 1. How much of 100 and the alumina becomes dissolved in the excess of caustic soda. If an excess of lime is used, the alumina will be precipitated, leaving the caustic soda alone in the solution. For making soap from aluminate of "Comparative Physiognomy." 2. How are rings soda about equal parts of lard and tallow are preferred, (finger) made? A. Finger rings are cast in moulds if and these should not be heated to a greater extent than is just necessary to liquefy them. The materials are the ends hammered or rolled down to the proper size not boiled in the usual way, but the combination is and bent into a ring around a mandrel; the ends are effected at the lowest temperature at which they can be then cut and fitted for the desired size of ring and solintimately mixed. Hot air engines are described in the dered together, then the rings are filed up and polished. SCIENTIFIC AMERICAN SUPPLEMENTS Nos. 162,247, 284, and 368.
 - be economy in employing the waste steam, as well also as direct steam in coils of iron pipe in a drying room. floor of the drying room, making the floor itself of large room flues. The slabs of scrap may be laid upon shelves. The coils of iron pipe may be made by any pipe fitting establishment.
- (51) W. J. D. asks: Who was the builder of the first locomotive, and where first used? A. Cugnot, a Frenchman, made a small locomotive in 1769, which is still preserved in the Museum of Artsand Metiers, Paris, Watt took out patents from 1769 to 1784, on steam carriages or wagons, but not known to have made any. Symington made a model of a steam carriage in Edinburgh, in 1770. William Murdock built and actuated a that the calks are red hot, either over a fire or in red hot locomotive in 1784, in Cornwall England. Oliver Evans, lead, dip the calks in a pan of powdered prussiate of of Philadelphia, obtained patents in Maryland, U. S., potash, and throw them into water. in 1787, for the exclusive right to operate steam wagon on roads and railways. In 1803, a Mr. Fredericks built a locomotive for a mine in Hanover, Germany. Trevitomatoes cause cancer is utterly without foundation. thick's first locomotive was running in 1802, and is con-There is not the slightest ground for fear in using freely | sidered the first effective effort on rails. Blenkinsop's locomotive, 1811. Hedley's locomotive, 1813, the the same time, they ought to be used like everything "Puffing Billy." Dodds and Stephenson commenced
- thereby derange the functions of the stomach. The and most powerful battery for electro magnet? A. The (52) W. J. D. asks: 1, What is the best acid nature of the fruit would, in such a case, perhaps, plunging bichromate battery is the best for this purpose. 2. About how large should an electro magnet be, and how much wire in the coils, and what size, to lift a weight of 1,000 pounds, or as much as it will lift conveniently, at a distance of from 1 to 3 inches? A. An axial magnet would be better for your purpose than the ordinary form of electro magnet. The size of the helix and of the wire of which it is made will depend upon the kind of battery used and the manner in which it is connected up. It would be more or less a matter of experiment to determine the size of coil required to lift 1,000 pounds from 1 to 8 inches. 3. How is an electric reservoir made, and about how long would the electricity last in a reservoir 8 feet long, 4 feet high and 4 feet wide-8x4x4 feet-to lift the aforementioned weight at every second interval by disconnection of the wires? A. For information on storage batteries, consult SUPPLEMENT. Nos. 304, 332, 370, 215, and 354.
- (53) J. H. K. asks (1) whether there is any metal or composition ("Delta" metal for instance) that is required of ends of a sewing machine bobbin. Would

- do not know whether their power depends on the length batteries. What book or books would best teach me all these things? A. The power of the current on the line depends on the number of cells of battery employed; the gravity battery is probably best for your purpose.
- (54) S. A. H. writes: In your paper August 10, in No. 4 of Notes and Queries, you state that a carbuteter for gas machine to supply 8 five foot burners should have 12 feet evaporating surface. I use Sisal hemp inmy carbureting tank, the air passing overthe fibers of the hemp, which have become saturated with gasoline; I therefore cannot tell what amount of surface I have. I use an iron tank 2 feet diameter and 2 feet high, fill it with hemp, and keep about 3 inches depth of gasoline on bottom; the air is introduced monia citrate of iron in 1 ounce distilled water. Also through a pipe, the end of which is under the surface of the gasoline; this is to supply five 6 foot burners; tilled water. The foregoing solutions are preparedsepa- have carbureter in a cellar the temperature in rately and kept from the light. 2. Chemical reaction which is sometimes as low as 32; use 88 gasoin making making blue prints? A. The ferric compound line. When weather is cold the gas is poor, making (ammonia citrate of iron) is by the action of the light bluelight. Is this the best way of making a carbureter reduced to the ferrous condition, which with potassium If not, will you please give directions for construction of a good one? A. Your carbureter is probably too small for cold weather. The iron tankin so compact a round ones. Any means that you can use for warmas to construction of pipe from the fan to the bench, (citric or hydrochloric) will produce the effect deing the air used in making the gas by taking it from say four feet. A. Wedo not think you will find it possessible by you, as it dissolves out the superfluous blue the cellar will materially help the process. Some use hot water in a pan under the carbureter in winter. There are a great many patents on carbureters; you could not do better than to make them a study. Copies will cost 25 cents each. 2 Is chrome steel much better for lathe tools, etc., than the ordinary cast steel? Does it require different treatment, in forging and tempering, from cast steel? A. Chrome steel is good, but we do not know that it is any better than the best tool steel. The treatment in forging and tempering is the same as for
 - (55) H. W. M. writes: 1. I would like to in Notesand Queries of a goodwork on physiognomy? A. To readfaces easily is in a great degree a matter of personal faculty not easily learned. There is an excel-lent book called the "New Physiognomy," also one on heavy, or rolled out in bars, cut off the proper length.
- (56) C. P. K. writes that during the operation of a steam fire engine near his bouse, an iron (50) C. W. H. asks: Which is the best leader shook very hard, so much so he was afraid it method of drying fish scrap after leaving a hydraulic would break, seeming to shake in unison with piston press, whether by some machine or by a kiln? A. rod of the engine. A. Probably the motion of the pis-When steam is used for boiling and pressing, there will, ton of the fire engine became synchronous with the vibrating properties of the leader, which induces vibration. This is a well known phenomenon in connection parts of hot water. This ink writes a brilliant bluish bitartrate of potash take the place of lime carbonate Another plan is to make flues in brickwork under the with bridges, which are often thrown into a severe vibration by a synchronous step or even the irot of a dog. tile laid over the flues, and carry the hot gases from the 2. In your issue of July 12, you speak of the Payne proboiler furnace through these flues to the chimney, or, cess for preserving timber. Can you tell me what that we do not think such a reaction would follow if the if not convenient, use a separate fire for the drying process is? A. Payne's process for preserving timber consists in impregnating the wood, while in a vacuum, with a strong solution of sulphate of iron, and afterward forcing into the timber a solution of sulphate of lime, or any of the alkaline carbonates, such as carbonate of soda, by which means the oxide of iron becomes insoluble.
 - (57) J. W. S. asks: How can I harden the calks on horseshoes of malleable cast iron or of steel? A. The steel is low, or decarbonized, and like the iron is amenable best to casehardening. Heat the shoes so

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5	Sewing machine, buttonhole, C. M. Banks Sewing machine, buttonhole, D. Mills	305,65
3	Sewing machine lamp bracket, A. Hartong	305,51
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	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305,595 305,433 305,575 305,683 305,471 305.727 305,703 305.489 306,500 306,420
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305,595 305,433 305,575 305,683 305,471 305.727 305,703 305.489 306,500 306,420
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305,595 305,433 305,575 305,683 305,471 305.727 305,703 305.489 306,500 306,420
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 305,433 305.575 305.683 305.727 305.723 305.489 306.500 306,420 305,495
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	Water and gas pipes, automatic electric shut-off for, T. P. Hughes Water apparatus for indicating the velocity or measuring the flow of, G. F. Deacon Water motor, J. H. Culver Weather strip, D. S. Early Weather strip, E. W. Philbrook Wheel. See Band wheel. Car wheel. Paper wheel. Wheel, S. T. Williams Wheel or pulley, E. B. Martindale window, W. C. Tucker Window bead fastener, W. H. H. Barton Wire stretcher, F. H. Bissell Wrench and pipe tongs, combined, D. Witt DESIGNS. Clo hes drier, G. W. Rhines Hammer or hatchet, claw, N. B. Allyn Ornamental stand, Cooper & Cohen.	305.484 305.502 305,595 305,433 305,575 305,683 305,773 305,723 305,733 305,433 305,433 305,435 15,427 15,422
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	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 305.433 305.575 305.683 305.471 305.727 305.703 305.489 305.493 305.495 15.427 15.422 15.423 15.423
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 905.433 305.575 305.683 305.727 305.703 305.489 306.500 306.420 305.495 15.422 15.422 15.422 15.422
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305,595 905,433 305,575 305,683 305,727 305,703 305,493 15,422 15,422 15,423 15,423 15,424 15,425
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 905,433 305.575 305,688 305,471 305.727 305,703 305.489 305.500 305,495 15,422 15,423 15,424 15,425
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305,595 805,433 305,575 305,683 305,727 305,703 305,493 15,422 15,422 15,423 15,424 15,425 11,510
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 905,433 305.575 305,688 305.471 305.727 305,703 305.489 305.495 305.495 305.427 15.422 15.422 15.423 15.424 15.425 11.500 11.516
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	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 905.433 305.575 305.688 305.471 305.727 305.703 305.489 306.500 306.420 305.495 15.422 15.423 15.424 15.425 11.506 11.516 11.507 11.509 11.508
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.494 305.502 305.595 905,433 305.575 305,688 305.727 305,703 305.493 305.493 15.427 15.422 15.423 15.424 15.425 11.504 11.516 11.516 11.517 11.509 11.508
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	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 805.433 305.575 805.688 305.471 305.7277 905.703 305.489 306.500 305.495 306.500 305.495 15.422 15.422 15.422 15.422 11.500 11.511 11.500 11.512 11.509 11.512 11.509
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 805.433 305.575 305.683 305.727 305.703 305.493 305.493 15.427 15.422 15.423 15.424 15.425 11.511 11.509 11.517 11.508 11.514 11.517 11.508 11.514 11.517 11.517 11.518
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 905,433 305.575 305,688 305.727 305,703 305.489 305,703 305.489 305,495 305,495 11,542 15,422 15,422 15,423 11,514 11,510 11,516 11,511 11,510 11,514 11,513 11,514 11,513 11,514
	Water and gas pipes, automatic electric shut-off for, T. P., Hughes	305.484 305.502 305.595 805.433 305.575 305.688 305.727 305.727 305.73 305.493 305.493 15.422 15.422 15.423 11.500 11.516 11.511 11.500 11.517 11.502 11.517 11.503 11.514 11.513 11.514 11.513 11.514 11.515 11.514 11.515 11.514 11.515 11.516 11.517 11.518
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	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305,595 805,433 305,575 305,683 305,727 305,703 305,493 305,493 15,422 15,422 15,422 15,424 11,511 11,510 11,511 11,500 11,514 11,517 11,500 11,514 11,517 11,500 11,516 11,511 11,517 11,500 11,516 11,517 11,500 11,517 11,510
	Water and gas pipes, automatic electric shut-off for, T. P. Hughes	305.484 305.502 305.595 905.433 305.575 305.688 305.727 305.703 305.493 305.493 15.427 15.422 15.428 15.428 11.500 11.516 11.511 11.500 11.512 11.513 11.501 11.513 11.501 11.516 11.511 11.500 11.516

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