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Steam Engines, Automatic and Slide Valve; also Boil ers. Woodbury, Booth \& Pryor, Rochester, N. Y. See

## NEW BOOKS AND PUBLICATIONS.

## An Illustrated Dictionary of Scientific

 Terms. By William Rossiter. NewYork: G. P. Putnam's Sons. 12mo, pp. 352. Price \$1.75.

A handy book of reference, containing some 14,000 sientific terms, many of them not to be found in ordinary dictionaries. The list includes the more important technical and scientific words, and those most comand usually the derivation. There have been added to the dictionary proper a number of tables of weights and measures, and brietly'the nomenclatureof botanical, zoological, anthropological, chemical, and geologi lassification. The illustrations are unimportant.
Reports to the St. Lojis Medical, Societt on Yellow Fever. By Wm. Hutson
Ford, A.M., M.D. St. Louis: Geo. O. Ford, A.M., M.D. St. Loujs.
Rumbold \& Co .8 vo pp. 320.
Ewbraces the report of the committee appointed by the St. Louis Medical Society to inquire into the rela-
tions of the epidemic of 1878 to the city of St. Louis, and Dr. Ford's Report on the meteorological conditions and etiology of yellow fever, on the etiology of suntroke, cholera, and other diseases associated with high emperature, and on the
The Building Materials of Otago and
Sodth New Zealand Generally. By
W. N. Blair. Dunedin, New Zealand:
J. Wilkie \& Co. pp. 244.

A volume of great local value and of considerable general interest, describing the building stones and croppings; the localities of the clay banks suitable for oroppings; the localities of the clay banks suitable for oricks, etc., and the qualities of such clays; New Zea-
and limes, cements, and their aggregates; and a verg interesting review of the numerous timber trees and
woods suitable for builders' use. There is added a section on the metallic products of Otago. The book is well indered.
Electro Metallubgy, Practically
Treated. By Alexander Watt. New
Treated. By Alexander Watt.
York: D. Van Nostrand. pp. 196.
This is the sisth and enlarged edition of Watt's handbook, from the English plates of 1876. A copious index

A Sketch of Dickinson College. By Chas. F. Himes, Ph.D. Harrisburg:
An interesting history of oneof the oldest colleges in the United States, neatly printed and illustrated by chapter is that tracing the progress in scientific educaon since the founding of che collego in 1783 ,
Seeing and Thinking. By William King.

dom Clifford. London: Macmillan \& | dom Clifford. |
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This, the satest voliveme of the Nature Series, includes our lectures by the late professor of applied mathematics and mechanics in Oniversity College, London, on the eye and the brain, the eye and seeing, the brain ad thlnking, and boundaries in general. No one who ever made an acquaintance with Mr. Clifford as a clear
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sented the ideal scientific intellect, at once earnest, sented the ideal scientific inte
fearless, and admirably sincere.
ndex to the Literature of Titanidm.
1783 to 1876 . By Edward J. Hallock,
8vo, paper. pp. 76 . Price 25 cents.
A paper read before the New York Lyceum of Natural History in 1876, and reprinted from the annals of the as that of Dr. Bolton's Indices to the Literature of ranium and Manganese.

Horse. By B. Pitcher. Second edition.
Chicago: published for
t published for the author.
hoeing, doctoring, and general treatment of the horse syong, who frankly declares himself to be no professor college graduate, horse doctor, or doctor of any kind;
but a humble mechanic. Mr. Pitcher is a practical but a humble mechanic. Mr. Pitcher is a practical
smith of nearly forty years'experienceand observation:
and he treats his subject wholly from the practical
standpoint. He has added a chapter of advice to standpoint. He has added a chapter of advice to youn no little practical wisdom and an abundance of sterling good sense.
First Step in Chemical Principies. By
Henry Leffmann, M.D. Philadelphia: Edward Stern \& Co.
Designed to make clear by explanation and illustration those points in chemical theory, notation, and Leffmann is the lecturer on toxicology at Jefferson Medical College, and his little handbook contains the substance of the lectures to the quiz classes of that in stitution.

## Min 1

HINTS TO CORRESPONDENTS.
No attention will be paid to communications unless accompanied with the full name and address of the writer.

## given to inquirers.

We renew our request that correspondents, in referring of of the question.
Correspondents whose inquiries do not appear after a reasonable time should repeat them. If not then pub-
lished, they may conclude that, for good reasons, the Editor declines them.
Persons desiring special information which is purely of a personal character, and not of general interest,
should remit from $\$ 1$ to $\$ 5$, according to the subject as we cannot be expected to apend time and labor obtain such information without remuneration. Any numbers of the Scientific American Supple-
ment referred to in these columns may be had at this ofice. Price 10 cents each.
(1) C. M. asks for a preparation to prevent nets from rotting in water. Mynets aremade of Ameri can hemp line, about $1 / /$ inch in diameter, and therefore
expensive. If $I$ knew of some preparation expensive. If I knew of some preparation or water proof misture that would prevent the water soaking
into them and rotting them, it would be a great saving tome. A. The following treatment is said to presery nets for a long time in good condition: Soften one lb. goodglue in cold water, then dissolve it in ten gallons of hot soft water with one half lb. curd soap. Wash the nets in soft water, then bo:l them in this for 2
hours, press out excess of the liquid and hang up ove night. The second bath consists of alum 21 l. ., water, gallons; heat nearly to boiling, and immerse the net a strong decoction of oak bark or a solution of sumas in warm water (water 5 gallons, sumac 8 lb .) and let them remain immersed in this for 48 hours, or longer f convenient.
(2) R. G. B. asks for a method for electroplatingflowers. A. See pp. 380.(39), 47, and34, Vol. 35. (3) L. C. P. asks: 1. What is a good wash (ed) for brick pathways? A. Try the following: red cher, 5 lb .; water glass, $1 / 2 \mathrm{lb}$. ; dissolve the iatter in Apply with a stiff brush, preferably whilehot. 2. What is a good remedy to vestroy ants in pantries, cellars,
(4) S. A. writes: I light my hotel with floor and above it the light is much more brilliant floor and above it the light is much more brillian of burners. Please explain the cause. A. In a column of air saturated with vapor of naphtha diffusion is never perfect, the heavier hydrocarbons tending to accumulate in the lower portions, and the combustion at these
points is incomplete in ordinary burners, owing tc the richness of the vapor in hydrocarbons.
(5) P. B. asks for a receipt for making a cheap imitation of mildew bronze. A. Dissolve equal weights of nitrate of iron and hyposulphite of soda in 8 parts of water; immerse the articles in this
until of the right tint, then well wash with water, dry, and brush; 1 part chloride of iron and 2 parts water im
parts to brass a fine antique green. Brush weil and lacquer with pale gold lacquer, or polish with oil
(6) G. writes: 1. On page 218 you publish an article on making cloth, etc., fireproof; would either
the first, second, or third composition also tend to renler the fabric mildew-proof: and if so, to what extent? A. Under ordinary circumstances the treatment would
prevent mildew. 2. Would frequent wetting and rough usage be apt to wash or shake off the.composition? A.
(7) N. C. M. asks for a recipe for making boots waterproof. A. Linseed oil, 1 part; mutton tal
low, $1 / 2 \mathrm{l}$.; beeswax $1 / 2 \mathrm{lb}$; melt and mix thoroughly together, and apply to the warm, dry leather with a orush. A small quantity of ivory black is sometimes added to this misture.
(8) H. K. \& W. M. W. ask (1) for a receipt or a size to mix bronze powder with so that it
can be painted on with a brush (on iron gas fixtures etc.). A. To one pint of methylated finish add 4 oz . of gum shellac and $1 / 2 \mathrm{oz}$. gum benzoin. Put the bottle in a warm place and agitate it occasionally. When the
gums are dissolved, let it stand in a cool place 2 or 3 days to settle, pour off the clear portion and reserve for fincst work, using the sediment, which by addition of more alcohol, may be made workable, when strained for first coat or coarser work. Add the bronze (q. 8.$)$
to this, and apply to the clean, smooth, warm iron, using a soft brush. Repeat, after drying, if necessary Thin with alcohol if necessary), to a aooid wrinkles and
orush marks. Varnish over all. 2. How is the glossy black obtained that $I$ have seen on . How is the glossy deliers? A. What you refer to is probably enamel or
Japanese work.
(9) W. H. F. asks: 1. What kind of paper process of stereotyping? on tissue paper; second, soft unsized printing paper of not too short fiber; backing cartridge paper What is the paste made of that is used in the paper process of stereotyping? A. Good starch paste answers very well. 3. Is there such a machine known as an enraving machine, and by whom manufactured? A. Yes; everal of these have been described and illustrated in the back numbers of the Scientific American. 4. an terra alba (or white earth) be used in making moulds astings, but not often.
(10) C. A. R. asks: 1. What is the value per ton of chromate of iron ore which assays 40 ther Eastern cities? A. About $\$ 30$ per ton. It work or other Eastern cities? A. About $\$ 30$ per ton. It would
not be profitable to ship such an ore East. 2. Is there any cheap mode of estracting the chrome from the ore, which could be set up at the mine, to save transportaion? A. The neutral chromate (yellow) and dichronate (red) of potassium, sodium, or calcium (lime), are the only preparations made direct from the ore. The chrome iron ore, previously pulverized and cleansed, is mised with carbonate and nitrate of potassa, soda, or
lime, as the case may be, and roasted on the hearth of a lime, as the case may be, and roasted on the hearth of a
reverteratory furnace. The sintered mass atter coolng is ground up, lixiviated with boiling water, silica, and alumina, precipitated from the solution by addition of wood vinegar (pyroligneous acid),and the clear liquid rawn off and evaporated until a film of saline material begins to form on the surface, when it is left to crysallize. From these crystals (yellow orneutral chromate of potassa or sodium), potassium (or sodium) dichronate is prepared by the addition to their solution of sul huric artalizes out on
(11) E. P. S. asks: Are there southern ghts at the south pole, as there are northern lights at ne north pole? A. Yes; Aurora Australis.
(12) C. J. D. asks (1) for a receipt for violin arnish (the Dest, if you please). A. Coarsely powdered am copal and glass, each 4 oz.: alcohol, 64 o. p, 1 pint; tirring, so that the bubbles may be counted as they rise until solution is complete, and when cold decant the
clear portion. When oil varnish is used it is made clear portion. When oil varnish is used it is made
from artists' vinegar copal. 2. Receipts for stains for from artists' vinegar copal. 2. Receipts for stains for
violins? A. To darken the wood rub over it nitric acid, specific gravity $1 \cdot 2$, and, after standing twelve hours, pecific gravity $1 \Omega$, and, after standing twelve hours,
wash and dry thoroughly. Then use either of the following: First, prepare a groundwork with strong bot aqueous solution of logwood extract; then apply a solution of 3 oz . potash, 3 oz . red sanders; $21 / 2 \mathrm{lb}$. gum shellac, and 1 gailon water, dissolved over a quick fire. 2. Boil 1 oz . logwood estract in 1 pint water (soft), and add $1-5$ oz. cream of tartar. Use the stain hot, and give several
coats, if necessary, drying between each. Use a " sawcoats, if neceessary, drying between each. Use a "saw-
edged " graining brush and asphaltum varnish, suffienty
(13) E. L. writes: We have tried a good manytimes, outwithout succes, to melt the following metals, namely, $1 / 4 \mathrm{oz}$. brass, 3 oz . pure silver, 1 oz . bismoth, 2 oz. common salt, 1 oz. of arsenic, and 1 oz. of potash. You will please let us know through your ScIntific Americais what kind of crucible to be used, crucible at once, how to melt it, and when melted hether to use plaster of Paris or brass moulds? A You can use a French clay melting pot (crucible). Melt he brass and silver first under a layer of charcoal, then dd your fiuses and finally the bismath and arsenic rapped in paper; after which pour as soon as possie, with care to avoid inhaling the poisonous arsenical fumes. With good management a four ounce charge nay be fused in a good ordinary stove. For larger can mould in plaster of Paris. See "Hints to corre pondents," above
(14) S. A. F.-The following is a good composition for blackboards: Shellac, 5 oz.; alcohol, $1 / 2$ pints; warm the latter and digest in it the shellac lter, and introduce about 5 ounces of a misture of equal arts bone black (floured) and emery four, stirring until uniform distribution of these is secured. The mirture should have the consistency of very thin sirup. Thin with more alcohol, if necessary, and apply two
(15) H. M. Co. ask: How can we in an in xpensive manner get the tin, solder, and dirt off old copper bottoms so as to make them clean? A. Cleanse frst in a boiling solution of 3 parts caustic soda, 1 part niter, and 5 parts water, and then in dilute sulphuric
acid; or dip momentarily in warm nitric acid, specific cid; or dip momentarily in warm nitric acid, specific
(16) H. L. W. asks: To what extent is air compressible? Or what is the limit of the compressi-
ility of air? A. We donot know that the limit of combility of air? A. We donot know that the limit of com-
(17) C. K. asks: Which chain has the most strength, one with $3 / 4$ twist links, or one with $3 / 4$
straight links? A. One with straight links, because straight links? A. One with straight links, because the stress is in the direct line
(18) C. N. K. asks if there is any way of nnding the number of pounds of coal that is necessary to run 1 horse power when you have the following
iven: 1. 'The sum of the horse power of each machine given: 1. The sum of the horse power of each machine.
2. The number of hours that each machine has run. 3 . The total number of pounds of coal consumed. Is ther any different way of arriving at the same thing? My object is to find out, after using several kinds of coal, duced by the combustion A. 1 horse power has been prothe other hand, with badly designed engines and boilers, 8 or 9 lb . of coal per hour have been consumed to produce the same result; the quality of the coal affects
the economy. The only way to get at the result you wish, is to weigh the coal consumed in a given time, wish, is to weigh the coal consume
and measure the water evaporated.

