

into the boiler shell, there is often corrosion caused by galvanic action, there being the essentials of a galvanic series—an attacking fluid, and two metals unequally attacked by a fluid. This action is intensified by the heat, and by any leak which may exist.

All these troubles exist on the outside—that is, the “fire” side of the boiler. Inside, the influences are more complicated, mysterious, and serious; but reason and experiment will baffle them.

Now sea water corrodes iron and steel plates quite rapidly, dissolving in a month 105.31 grammes of steel from a plate 40 centimeters square; and in the same time 99.30 grammes from an iron plate of the same size. Iron kept in water containing carbonic acid gas oxidizes rapidly with escape of hydrogen gas, proving the decomposition of the water, apparently by galvanic action, or rather by what is called catalysis, where one element, not itself attacking another, causes a third to do so. Dry oxygen does not corrode bright steel or iron; damp oxygen slightly corrodes them. Dry carbonic acid has no action thereon; damp carbonic acid forms a white carbonate of iron on them. Dry carbonic acid and oxygen have no effect, while damp carbonic acid and oxygen have a very rapid oxidizing action.\*

Distilled water, free from air or gases, does not corrode iron, it being very difficult to get a bright blade immersed therein to do much more than slightly spot with rust; and careful examination of these spots generally shows at each point an impurity in the iron sufficient to induce a galvanic current, just as a piece of zinc or copper placed against the iron would do. Trying lead plates, it is found that while distilled water free from air eats off in two weeks, from a square meter of surface, only 1.829 gramme, the same quantity of the same water aerated dissolved away 110.003 grammes.†

The presence of chlorides of magnesium, ammonium, sodium, potassium, barium, and calcium dissolved in water largely increases its rusting action on iron. The magnesium chloride is the most active of any one of these; but in conjunction with lime carbonate is also active; as are mixtures of the calcium chloride with that of sodium or of barium.

The chloride of magnesia solution is of all these, however, about the only one that attacks iron at 212° Fah. when there is no air present.

Considerable trouble is often caused where the injection condenser is used, and the condensed water contains slight quantities of lime and magnesia salts, which, at say 150° Fah., form soaps with the grease brought over from the cylinders, etc. At higher heats these soaps decompose into free fat acid (generally oleic), and a basic lime soap, which at still higher temperatures may be carbonized. The soap adheres to the boiler surfaces, and the acid attacks the iron, which darkens the scale.

Even if there be no salts brought over, the destructive distillation of fatty matters is, while giving no scale, none the less injurious and destructive than in the case last cited.

Where the water contains lime and magnesia salts and fat acids, the remedy is lime water and caustic soda, which remove both the fat acids and the magnesia.

There are so many cases where boilers fed with “pure natural water” have been rapidly corroded away, that steam users congratulating themselves that they are free from the evils of scale should see if they are not using pure water containing gas in solution, and if there be this trouble, it may be cured by a regular dose of whitewash, or by mixing calcareous water with the soft gas-holding water.

When pure distilled water is used there will be no contained gas and should be little trouble from corrosion. Perhaps for marine purposes it will be impossible to escape corrosion without employing copper boilers, and even then we are not so sure about it.

**SWISS TESTIMONY TO THE ADVANTAGES OF OUR PATENT SYSTEM.**

Hitherto anti-patent men have found their strongest argument against the recognition of any property right in inventions in the practice of Switzerland. “Here,” they have said, “we see the benefits of free trade in ideas. Switzerland wisely refuses to allow her industries to be taxed and overridden by patent monopolists. See how prosperous she is—how successful her manufactories—how skillful her artisans! Be wise and profit by her example.”

At first thought, nothing would seem more reasonable than to suppose that a manufacturing country which should reserve to itself the right to appropriate the inventions of all nations without payment of inventors’ fees would be so much in pocket, at least. But the experience of Switzerland, where the experiment has been tried under the most favorable conditions possible, does not make the supposition good. On the contrary it has proved decidedly a losing game; and the loss has fallen where it could least be afforded—on the industrial character and productive capacity of her artisans. The Swiss workman has dropped behind in the contest for mastery, and Switzerland’s trade is departing in consequence.

Take, for example, the shoe trade. The largest shoe factory in Europe is at Shoenwerth, between Bâle and Zurich. It was set up for its owner, Mr. Bally (one of the Swiss Commissioners to the Philadelphia Exhibition), by American

mechanics; and it is stocked with the best American machinery. Mr. Bally is a man of exceptional force and business ability. He has visited this country often, and is familiar with American methods of organizing labor. He is careful to secure promptly every new invention bearing on his business. He has no royalties to pay; and he pays his workmen less than American rates. Yet he cannot compete with New England, even in Swiss markets. He has lately recounted his experience in this connection in a pamphlet addressed to Swiss manufacturers; and he traces the inability of his workmen to compete with Americans to their inferior intelligence and skill, an inferiority mainly due, he is quite sure, to the lack of the stimulus of a patent system. He tells his countrymen very frankly that their industries are seriously overshadowed by those of America, and that their industrial salvation must be looked for, largely, if not mainly, in a patent system approximating ours. He says: “We must introduce the patent system. All our production is more or less a simple copy. The inventor has no profit to expect from his invention, no matter how useful it may be. On the contrary, each one has the right with us to appropriate to himself an invention, to copy it, to the great injury of the inventor. It is evident that this absolute want of protection will never awaken in a people the spirit of invention, but on the contrary it accustoms them more and more to copy that which belongs to their neighbors, and that is not to the honor of our country. The want of protection for new inventions is a great disadvantage to us. The State ought not to hesitate to add to its resources this new resource. But at the same time we must remember that an invention is valuable in proportion to the facility with which it can be made available, and so it is essential that the grant of patents be accessible to inventors of the most moderate fortunes.”

In an appendix to a French edition of this pamphlet, Mr. Edward Dubied, from the standpoint of the watch manufacturer, quite astrenuously insists on the immediate adoption of a good patent law. After reviewing several lines of production in which American competition has brought things to a desperate pass in Switzerland, he says:

“At this rate, there is no reason why all our industries should not be overwhelmed, one after another, by those of America; and yet, when we ask what wages are paid the workmen in the latter country, we learn with surprise that they are three times as much as those which our workmen, both artisans and farm hands, receive. The conclusion from these facts is that our intelligence and productive power, compared with those of America, are as one to four—a proportion which we must admit, if it is true that an American factory which pays its workmen three times as much as a Swiss factory, and has to give a much higher rate of interest for its capital, nevertheless can produce at less cost.”

Two things are requisite, Mr. Dubied goes on to say, to get them out of the plight they are in. First, a good patent law; and second, an increase of the technical instruction of their artisans, foremen, and masters. He says: “Our readers are perhaps astonished that we insist upon a patent system as of the first necessity; but we shall justify this by proving that the protection of property in inventions develops the desire for technical instruction, while the absence of such legal protection is nothing more or less than a premium given to ignorance, to the detriment of inventive talent.” Further on, he points out the secret of the educative influence of patent rights by showing that in patent granting countries intelligence, technical instruction, and inventive intellect have a real value.

Mr. Dubied’s testimony is so strong and so much to the point, withal so pertinent to the discussion in progress here, that we cannot refrain from quoting his final words in this connection: “Messrs. Favreperret, Bally, and David, our Commissioners to the Philadelphia Exhibition,” he says, “call for a patent law in Switzerland as a means for perfecting our industries. The author of these lines regards the institution of patents as the first and indispensable measure, without which any other will be utterly useless, for reaching the end that we all have in view. If he especially insists upon this point it is because he had the advantage over the gentlemen he has named, of spending twenty-five years as engineer and machine builder in a patent granting country—namely, France—before he established himself as a manufacturer in Switzerland. He can, therefore, bring his own experience to the support of their demand; and he assures his fellow citizens that a law for the protection of property in inventions would be a true magician’s wand among us, completely transforming our system of manufactures, and raising us in a short time, in a natural manner, and with less effort than we should expect, to a level with the nations most advanced in the arts. . . . Away with those false principles which conduct an industry to certain ruin. Let us delay not a moment to obtain a good patent law.”

We would respectfully commend these expressions of dearly bought wisdom to those gentlemen at Washington who are dallying with “those false principles which conduct an industry to certain ruin.” The most enlightened minds of the most enlightened countries are convinced that the prime secret of American superiority in the industrial arts is due to a patent system, the inspiring, educating, and encouraging influence of which reaches every grade of society. Thus far it has been conducted with a view solely to the advancement of the arts through the encouragement of inventors. To “amend” it, as now proposed, so as to make the inventor the cat’s paw of the infringer would be to cut the

very heart out of the system, and put a summary check to our industrial progress.

**A REMARKABLE PICTURE OF THE MOON.**

There are perhaps few persons who, in passing up and down Broadway during the last few weeks, have not had their attention attracted to a remarkable and strikingly brilliant picture of the crescent moon exhibited in the show window of Messrs. Scribner, Armstrong & Co.’s book store. It was a happy thought that led Mr. Henry Harrison to attempt this painting, and the success that has crowned his efforts affords a most excellent example of the results that one may attain in such matters, when to the skill required for manipulation is joined an absorbing love for the object of representation as a subject of study. For it must be stated that Mr. Harrison is an astronomer; and while he has displayed in his painting all the sentiment and all the technical skill of the artist, that “high art” feeling which prompts the belief that “it is not the mission of art to represent nature, but only to use her as a means to express an ideal,” he has subordinated to scientific accuracy; and herein lies the great interest and great value of his work. So, with a knowledge of the artist’s motives and of the means that he employed to secure accuracy in the measurement of distances, and in the colors and contours of the objects presented in the lunar landscape, we can scarcely be impressed by any other feeling in looking at this canvas than that we are gazing, not at a mere picture, but at a reality—at the wildly desolate surface of the satellite as she might appear to us could she be brought within range of our unaided vision.

The canvas is unpretentious in size, being only 27x27 inches; the painting represents the moon about three and a half days old—i. e. “in her crescent”—with the terminator at Mt. Glacier, the edge toward the sun bathed in most brilliant sunlight, shading off into a light yellowish tinge, and then blending into the darkness of night toward the terminator. In the earthshine, or surface in shadow, may be seen some of the most prominent features, such as the craters *Copernicus* and *Tycho*, the *Apennine Mountains*, and nearly all of the *Meres*. The whole orb stands out in bold relief, against a dark sky blue background, the exact color of the field of the telescope an hour after sunset.

The moon has been a subject of topographical and pictorial representation by astronomers for ages past. Its entire surface has been surveyed and mapped in outline, more or less accurately, by Lohrmann, Herelius, Baer and Maedler, and Schmidt; drawings of single craters and casts of the whole planet have been executed by others, and the development of the photographic art has been the means of production, by Messrs. Rutherford and Draper, of lunar pictures nearly absolutely correct.

Yet, if we except some small water color sketches of some of the more prominent mountains and craters, reproduced in print to illustrate Neison’s work on “The Moon,” and a few others by the Astronomical Society in London, we believe that Mr. Harrison’s is the first attempt to render a faithful picture in colors of the moon as it appears to us in the telescope, showing its delicate gradations of light and shade, its enormous circular caverns or pits strewn with boulders, its level plains, its brilliantly illuminated towering peaks and crater walls, its ever varying terminator, and, above all, that lustrous sheen that is all its own, and that has made it recognized as the “silvery” planet. It is Mr. Harrison’s intention to publish, in oil color chromos, a series of six *facsimile* reproductions of paintings of the moon in its progressive phases from the “three days’ old crescent” (just noticed) to the “full moon” and “last quarter.” We see no reason why (if the reproductions come up to the standard of excellence shown in the original) the venture should not prove a perfect success, through generous aid accorded him by all who are interested in the advancement of science and art.

**American Petroleum Exports.**

The exports this year have been larger than for any year previous to 1877, the total exports in gallons from January 1 to May 11 having been for five years:

1878.	1877.	1876.	1875.	1874.
76,623,252	87,252,268	72,024,491	60,542,620	71,176,609

Before 1874 the exports had never reached 60,000,000 for this period.

The distribution of the exports from the different ports is a matter of considerable interest, as it is now supposed to be substantially regulated by the contract of the Standard Oil Company with the railroad companies. Last year, it will be remembered, the proportion exported from New York increased enormously, largely at the expense of Philadelphia. This was chiefly due to a contest between the Standard Oil Company and the Pennsylvania Railroad, by which the former, controlling most of the petroleum to be shipped, refused to send anything over that railroad. That conflict broke out just about a year ago and lasted six months. Thus the part of the years for which the above figures are given was uninfluenced by this contest. New York exported 71.3 per cent of the total both years; Philadelphia, 15.2 per cent last year and 13.8 per cent this; Baltimore 9.8 per cent last year and 13.4 per cent this. New York has never, or at least not for several years, exported a larger proportion than this year; Philadelphia, on the other hand, has never exported a smaller proportion (28 per cent in 1876 and 17.2 in 1875); Baltimore, in spite of its increase, has not this year reached the proportion which it reached in 1876 (16.7 per cent). So far this year the reports show that 71.3 per cent of the whole has gone by way of New York, 13.9 by Philadelphia, 13.4 by Baltimore, and 1.4 by Boston.

\* The reader will see that the influence of dampness, etc., in air is of importance as regards the corrosion of parts of iron railway bridges, and other similar structures, especially where not well painted.

† It will be seen that this has a bearing upon the water pipe question, but we will not discuss that now.