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## \%aticer Milunties

A. J. will find directions for stereotyping
on p. 363, vol. 30.-J. T. can blue steel articles by the processgiven on p. 123, vol. 31.- J. K. will find an explanation of the theory of the crank on p .
112, vol. 31.-R. J. T. should consult a physician.J. F. is informed that etching on glass is described on p. 409, vol. 31
(1) W. B. asks: At what angle should mower knives be ground, so as to give the best
and most desirable edge? A. It depenas npon the material to be cut.
(2) W. K. says: Inasmuch as the question whether sparks of electricity can be discharged from the body of a workman in a powder mill, and
cause an explosion, is agitating the minde of sclentists, I send you an account of an incident which came under my observation a number o years ago. On a bitterly cold wintry day, a friend and myself started in a sleigh to make a visit. On retiring at night, my friend, after having removed an entirely woolen garment, threw it care-
lessly over his arm. Having laid it down, he aplessly over his arm. Having laid it down, he ap-
proached the said garment with his other hand; and upon so doing, he was amazed to see an immense number of sparks all over the garment, as if it were on fire. This was accompanied with a noise similar to that produced by a voltaic bat-
tery. He withdrew his hand, and the sparks were tery. He withdrew his hand, and the sparks were gone, then reapplied it, and the same phenomenon Reapplying his hand for the third time, but a few sparks were seen, without any noise whatever. Where did the sparks proceed from? A. The sparks were caused by friction. They are often observed
in removing woolen clothes in winter, especially in removing woolen cloth
when the air is very dry.
(3) R. B. B. asks: What would be the pracical difference between cutting off steam at $\frac{1}{8}$ th stroke, and throttling the steam until just $\frac{1}{8}$ of
the full working pressure is realized? the full working pressure is realized? A. To cut
otif the steam at $1 / 8$ stroke would be the more economical, there being in that case less loss from condensation.
(4) A. M. asks: How can I produce a fine high color on gold jewelry? A. Boill 8 ozs. salt-
peter, 4 ozs. alum, and 4 ozs. common salt together in a porcelain or other fireproof vessel (not metallic), in barely sufficient water to dissclve them; add 9 ozs. strong muriatic acid to this solu-
t:on, and filter. This quantity will be sufficient for tion, and filter. This quantity will be safficient for coloring 4 ozs. of work at a time, and should be
kept in a well stoppered glass bottle when not in kept in a well stoppered glass bottle when not in
use. A nother recipe is: Boil 1010 ozs. saltpeter and $53 / 4$ ozs. common salt together, in a porcelain dish, in a quantity of soft water barely sufficient to dissolve them, and $1 / 8$ oz. nitrate of silver (lunar cuustic) and $93 / 2$ ozs. muriatic acid; filter. This quantity will be sufficient for coloring 4 ozs. of work at a time, and should be kept in a well stoppered glass
bottle when not in use. To color with either of the foregoing mixtures, anneal the work twice and boil it, each time after annealing, in a pickle sonsisting of 8 parts water and 1 part sulphuric acid. Then pour a sufficient quantity of the coloring mixture into a porcelain dish, and heat it to about $150^{\circ}$ Fah. Hold the work in this for about two minutes, then take it out and rinse it in clean
water. If not sufficiently colored to suit, repeat the processuntil the desired color is obtained. Anis to make a kind of paste consisting of 2 part saltpeter and 1 part each of alum, sulphate of zinc common salt, and a little water. These ingredi ents are well mixed in a mortar, and the articles to be colored are covered with the paste, laid up-
on an iron plate, and heated over a clear fire nearly to a black heat. They are then suddenly plunged into cold water and well washed. This nsures a beautiful high color.
(5) F. H. D. asks: Will a locomotive, stand
ing still on an up grade, move or start as easily up ing still on an up grade, move or start as easily up
the grade when the crank is on a line from the axle to the point of the wheel which rests upon of rail as it would if the crank was at either side this line? A. A lscomotive should start equa is on the absolute dead center, the other is in a position to exert its maximum force.
(6) H. J. E. asks: 1. How much sulphuric acid must put to a gallon of water to take the
cale from castings that are to be used for stove patterns, and how long should they remain in the acid? A. About one of acid to fifteen to twenty of water. 2. What kind of a bath should the castings be put into after they are taken from the
acid? A. Wash with clean water, and keep in a quantity of water containing a little potash or soda in solution until required. 3. How should wax be applied for waxing iron stove patterns? A. The wax is applied by simple friction with the metallic surfaces. This increases the temperature sufficiently toimpart a fine even coating to the metal 4. How can I make good varnish for wood pat-
terns? A. Use shellac in alcohol. A little gum terns? A. Use shellac in alcohol.
(7) W. B. M: asks: How can engraving breaking. A. Use Stubs' steel, heat it to a low red heat, a nd temper to a straw color

1. What is the best way to keep a boiler when
not in use? I have one under my charge that is not in use? I have one under my charge that not used for several months in fall and winter. way to keep it? A. Yes. 2. I blow off a few
inches at a time, never blowing out entirely when hot. A good many around here blow their boilers
out while hot, and this of course has a tendency
to cake the slush, etc., held in solution, on the plates; at least that is my opinion. Am I not right? A. Yes.
(8) W. F. R. says : I claim that a powerful lamp for heating purposes will give off more heat
to the room if a piece of sheet iron be placed horto the room if a piece of sheet iron be placed hor-
izontally just above it than if nothing be placed above it. My friend says there will be no differgiven off by the lamp will be equal in both cases, but the metal would distribute the heat more evenly throughout the room.
(9) A. H. asks: How can I cement emery to mixed thickly; and when dry and hard, apply mixed thick1y;
emery and glue.
(10) T. D. asks: How can I obtain the in dex of an engine lathe? I have applied to some machinists, and I purchased screw cutting tables; and although I can strike some threads, still I am unable to ascertain thepropermethod,as my gears
all run differently from those described in any published system. My leading screw is an 8 thread $91,48,105,112$. The gear on the spindle (under the head stock on cone) is 46 . Can you give me any instructions in regard to filling out an index? Multiply the number of threads to the inch you Wish to cut by ony small number, for instance, 4 , Then multiply the number of threads to the inch in your lead screw by the same number (4), which will give you the gear wheel for the spindle. If thet, $\operatorname{tr} 2$, thus obtained are sizes you have not plier.
(11). P. says: A rubber belt slips very some say pitch. Will you give me rosin or it, and A. Use a mixture of equal parts of red lead,French ellow, and litharge ; mix with boiled linseed oil nd japan sufficient to make itdry quickly.
(12) K. B. says: We recently cast a pair
brass boxes for the crank wrist of an entine o 100 horse power. The boxes were finished at th made. After the engine had run about 20 minutes the boxes became so hot that it was necessary to until we were obliged to remove them. The me condemn the boxes on the ground of being mad of poor metal. Theboxes made in the proportion of 1 part block tin to 9 parts good scrap
copper. Do you consider this a good material? If so, what was the probable cause of their heat ing? A. Your mixture of brass was a good one
though 1 part less copper would have been bette The heating may have occurred from the brasse not leading true, from the wrist not being true, from its having too little wearing surface, from box, or from any one of a variety of simila causes.
(13) R. asks: 1. Where should I apply for place on the school ship Minnesota? A. On
board. 2. Will the authorities take a boy who is not a native? A. Yes.
(14) S. S. S. asks: What is the best treatabsorbent of water and easily kept clean? A Apply raw linseed oil, as follows : Take a block of vood, about 10 by 15 inches and 2 inches thick; tack woolen cloths upon it, saturate the cloth
with the oil, and rub the floor till ycu produce a
wall? A. Take paint for a smooth, hard-fnishe to a gray with a little black and yellow, using ver fitte of the latter, so as to give only a slight tint wall, appearing spotted in places; repeat the coats until there is a uniform $k$ loss throughout; this
(15) J. W. S. asks: How are twisted gun
barrels made? A. After the square bars are made, barrels made? A. After the square bars are made, hey aretwisted around a mandrel and th
d. See No. 1 of "Practical Mechanism."
(16) J. H. A. asks: 1. Will soft gray iron Yes. 2. Will they receive and discharge magnetA. Ye A. Yes. 3. Is Léclanché's battery a patent? A length of wire, a long slim coil or a short thic one? A.A thick wire with few convolutions give the best effect when the amount of current pass ing through it is comparatively great. A thin wir is small but has considerable electro-motive force 5. Is a cone of hourglass shape, with broad poles
any better or as good as a straight one? A. Somewhat better for equal weights.
(17) E. C. H. asks: 1. In making the core or for thepattern of a steam cylinder, should it be made exactly the same size as the core prints, Green sand cores shrink, while dry cores expand a little. The amount is, however, in either case too smallto render any allowance necessary. 2.I is, to produce the various articles of a steel difficulto forge? A. Yes.
(18) J. M. S. asks: 1. Will No. 16 coppered wire do to construct a telegraph line $1 / 2$ a mile in the best, a ground or a return wire? A. A return wire. 3. How many more cells of battery will it take with the ground than the return wire? A hree times. 4. Can a good ground be made by and then filling with scraps of old iron? A. Yes This
wet.

Minerals, mtc.-Specimens have been re eived from the following correspondents,and oramined, with the results stated
W. B. B.-Your boiler deposit consists of lime Iumina, oxide of uron, and oily matters.-B. B.

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Also inquiries and answers from the following:
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