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F. J. will find Professor Böttger's recipe for a depilatory on p. 107, vol. 30.-F. R. S. can make a colorless varnish by the process described on p. 150, vol. 29.-J. T. W. will find that the field camera is described on p. 58, vol. 31.-A. J. T. will find that induction coils are described on pp. 215, 218, 363, 378, 379, vol. 30.-R. N. will find a full description of the process of lithography on p. 298, vol. 31.

(1) W. H. says : I have found in California a deposit of almost pure silicate of alumina. It is entirely free from any metal, and contains about 2 per cent of silicate of magnesia. Is there any use for it? A. It could be used for filling the walls of fireproof safes; and if it could be easily handled and cut into blocks, it might be employed in the lining of furnaces, or in similar cases where great heat is encountered.

(2) P. I. O. asks: If a ball of iron, or other substance of greater specific gravity than water, be placed in a vessel (of sufficient strength) full of water, and submitted to powerful hydraulic pressure, would the pressure cause the water in the vessel to float the ball? A. Yes, if the water were sufficiently compressed. It would, however, require a machine of extraordinary strength, as a pressure of 1 atmosphere only increases the density of water about 0.0000033.

(3) W. H. R. asks: Will a mark on a tree indicating high water be permanent, or higher up by the growth of the tree? A. If the tree has attained its growth, the change in the position of the mark will be scarcely perceptible. In other eases it may change considerably. We have heard of "blazes" made on trees which were afterward found to be high above the reach of a "blazer" ' on the ground.

(4) J. T. H. asks: If a belt slips, does it slip on the driving pulley or the one driven? A.In most cases, but not always, on the driven pulley.

(5) H. C. S. says: I made a piece of leather hose 41/2 feet long and 21/2 inches in diameter, and put a water pressure upon it of 275 lbs, to the

(9) J. B. says: We use salt water in a steam boiler. We have no condenser, and do not know how to make one. We can get sulphurous water, but think that will be as hard on the boiler as salt water. What would you advise? A. It will be necessary for you to blow off frequently. so that the saturation of the water in the boiler does not become very high. Even with this pre-caution, scale will form in the boiler, and should e removed every few weeks.

(10) J. E. H. says: Some time ago I saw a recipe in the SCIENTIFIC AMERICAN for cleaning guns by using quicksilver. It stated that the quicksilver would form an amalgam with the lead in the barrel of the gun. I have tried it and have signally failed. I then took some shot and put in the mercury to see if it would form an amalgam; and the shot still remain in it as perfect as when first put in. What is wrong? A. If you will melt the shot you allude to, and draw off the clear metal from the oil and plumbago with which they are covered, you will have no difficulty in combining it with the mercury. For the same reason you will see that it is necessary to first remove all the carbon, etc., from the interior of the barrel of the gun before the required action will take place.

(11) L.G.P.asks: What kinds of gumscanbe dissolved in sulphuric ether? A. Most gums are soluble in ether.

1. How can I color collodion? A. Use any of the aniline colors. 2. What will dissolve collodion after it is dry? A. Ether containing a little alcohol.

(12) J. H. S. asks: 1. To how many grains (by weight) of best rifle powder is one grain of good gun cotton equal? A. One part by weight of gun cotton is equal in projectile power to 45 or 5 parts of gunpowder. 2. How is gun cotton manufactured for use in fire arms? A. The cotton is steeped in a mixture of equal parts of strong sulphuric acid, specific gravity=1.84, and fuming nitric acid for about 15 minutes. Gun cotton has not been adopted in practice as a good substitute for gunpowder; its large bulk, coupled with the fact that the explosion is attended with the evolution of much water and nitrous acid, render it inconvenient as a substitute for powder. 3. Is it more dangerous than gunnowder? A. Yes: gun cotton is liable to spontaneous combustion. In some cases it has been known to ignite at a tem perature of 43°. A small magazine filled with gun cotton in the Bois de Vincennes, Paris, was exploded by the sun's rays. The combustion of gun cotton, unlike gunpowder, takes place almost instantaneously, and for this reason it is liable to injure or rupture the fire arm in which it is used.

(13) F. W. B. asks: Are there any acids that will eat into or soften the surface of rocks? A. Most of the materials forming rocks are either partially or wholly soluble in acids, or mixtures of acids. They are not all soluble in the same acid. As you failed to state any particular variety of rock, wemust omit giving you here a complete dictionary of solubilities.

(14) B. H. W. asks: What substance is used to cement the seams of gas bags? A. For this purpose a cement used is obtained by dissolving gum rubber in hot naphtha, which is then allowed to evaporate until of the consistence of a thick paste. The seams are cemented with this, and placed under great pressure for some time. They are then (usually) countercovered by a strip of the same material as the bag. This is fastened on by means of the above cement, so as to form a binding. When this is finished, it is again placed under pressure. A bag, so prepared, will stand 600 lbs. pressure, and is generally warranted to stand 1,000 lbs.

(15) E. E. asks: What would be the best to fill an ice box between the outside and inside of the box? A. Charcoal.

(16) W.C. asks: Is there a chemical that will dissolve gutta percha, and will dry as fast as bisulphide of carbon? A. Bisulphide of carbon is by far the best solvent for gutta percha, we believe, thus far discovered; but if the gum be digested for some time in hot naphtha, a solution in that liquid may be obtained. But as naphtha, like carbon bisulphide, is an extremely volatile liquid, and when its vapor is mixed with air forms a terribly explosive mixture, it follows that, to obtain the desired solution without a prodigal waste of the solvent and its accompanying danger, it is necessary that the materials be inclosed in a strong, airtight vessel, and heated over a water bath so as to insure a uniform degree of temperature, which should not be allowed to exceed 100° Fah.: and even then we should not consider the above method free from danger in inexperienced hands.

(17) C. G. H. says: On April 13 I was carrying an umbrella in a snowstorm. I touched the frame of the umbrella and received a spark of electricity. On presenting my finger, I could plainly see a spark 1/6 of an inch long. What is the cause? A. Occasionally, during a snowstorm such as you mention, or when for some reason the air is extremely dry and the atmosphere becomes very highly electrified, like strange electrical phenome na often occur. Your umbrella, in this case, probably acted the part of Franklin's kite, and your body formed the conducting link to the earth. (18) E. W. P. asks: How can I make pow dered charcoal into a porous cake for use as a filter? A. One method consists in pulverizing animal charcoal until reduced to an impaluable powder. This is mixed with a definite proportion of Norwaytar, and a compound of other combustible substances. The combined materials are then properly amalgamated with liquid pitch, and the whole kneaded up into a homogeneous plastic mass which admits of being molded into slabs or These blocks, having been allowed to dry and mation can in this way be expeditiously obtained.

harden, are subsequently carbonized by being subjected to a process of incineration by heat; and in this manner all the combustible ingredients are burnt out, leaving nothing behind but the anima charcoal in the form of a block of charcoal, permeated throughout by innumerable pores.

(19) W. H. says: 1. You say that, by coating a room with oxide of zinc and glue size, and when dry, coating with a solution of chloride of zinc, can obtain a good gloss; must there be any glue in the chloride solution to bind it? A. It is not necessary. 2. What strength of solution of chloride is required? A. Use a strong solution. 3. Chloride of zinc is extremely dear here. Is there any substance that can be used to produce same re sults and be sheaper? A. We do not know of any substitute other than good varnish. 4. Does the above compound affect colors? A. Yes; some colors are apt to be partially bleached when mixed with it. 5. Is the chloride of zinc hard to make? A. No: dissolve metallic zinc or its oxide in a small quantity of hydrochloric acid until it will dissolve no more; this solution is then ready for use. When applied to the coating of zinc oxide, it combines with it to form a subchloride or oxychloride of zinc, which is neutral. 6. Is silicate of soda a strictly waterproof substance? A. It is not. 7. What can I add to its brittleness to keep it from becoming cracked and mealy when exposed to the air a short time? A. We do not know that you can better it at al!. 8. Could I use this mixture on plastered walls to keep water from staining through kalsomine? A. It might be made use of for this purpose, but we do not think it would be verylasting. 9. Would it peel off in damp places? A. Yes.

(20) E. C. S. asks: When it is exactly noon at any given point on a given meridian, is it exactly noon at every other point on the same meridian? A. Yes. 2. I read that the variation in the length of the solar day is owing to two distinct causes, the inequality of the earth's motion around the sun, and the inclination of its axis to the plane of its orbit; and that from these two causes arises the difference between sun time and clock time. I can see how the earth's unequal motion may make the solar day more or less than 24 hours: and I can see how the inclination of its axis may affect the time of sunrise and sunset, and the length of day and night; but I cannot see how it affects the length of the solar day or makes sun fast and slow. A. If the books state that the variation of the length of the solar day is due to the inclination of its axis to the plane of in orbit, they can only allude to the variation in the duration of daylight from summer to winter; and if they speak of the inequality of the earth's motion around the sun, they mean the yearly motion, during which the earth recedes in spring time from the sun, and retards her velocity, while in the fall she approaches the sun, and accelerates. The daily rotation is perfectly uniform, as proved by observations of the stars, which show that the sidereal days, which the length is only 23 hours and 56 minutes very nearly, never vary in length. The difference between solar time and the time of a well regulated clock would not exist if the earth's yearly orbitwere a circle, and her yearly motion uniform. This difference, which is commonly described in the almanacs as "sun fast" and "sun slow," is caused by the inequality in the yearly motion of the earth, by which the apparent motion of the sun along the signs of the zodiac (and which, on an average, is 59 minutes and 3 seconds of a degree per day) becomes retarded or accelerated : the difference of this apparent irregular motion, with that of a sun whose apparent motion is regular, constitutes the difference between the so-called solar time and mean time. For instance, from June to September, the earth recedes from the sun, and its velocity is retarded, and the sun's apparent motion along the zodiac is also retarded, so that the statement in the almanac, "sun slow," is correct.

COMMUNICATIONS RECEIVED.

The Editor of the SCIENTIFIC AMERICAN acknowledges, with much pleasure, the receipt of original papers and contributions upon the following subjects:

On Commercial Sponges. By S. H. T. On Flower Growing. By W. H. M. On the Tides. By A. McN. On Disasters at Sea. By J. F. J On the Locomotive. By J.F. J. On Railway Rails. By J. P. T.

Also enquiries and answers from the following . C. N. F.-S. T.-L. D. D.-A. W.-P. F. M.-C. N.F -F. W. J.-B. S. K. O.-C. F. A.-E. J. F.-W.L. D. -C. J. B.-C. S.-C. D. G.-S. G., Jr.-H. S.-J.P.M. -E. F. O.

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American Metaline Co., 61 Warren St., N.Y. City. Grindstones, 2,000 tuns stock. Mitchell, Phila., Pa. Small Tools and Gear Wheels for Models. List free. Goodnow & Wightman, 23 Cornhill, Boston, Mass.

engtn thingsbeing equal, stand the same pressure? A. Yes.

(6) W. J. F. asks: How many square feet of cooling surface are usually allowed per horse power in a surface condenser? A. It is a common rule to allow from 1/2 to 3/4 as much condensing surface as the heating surface of the boiler.

(7) G. S. C. asks: What makes some steam boilers, if the water is cold when the fire is started. give a clacking and hammering noise internally? A. It is no doubt due to sudden expansion of parts of the boiler, caused by the circulation of the water as it becomes heated.

(8) W. R. H. says I read that Lieutenant Hitchcock of the United States Coast Survey discovered what is apparently a spring of fresh water at sea. Are there any more localities in the ocean where fresh water can be obtained? A. Fresh springs in salt water are by no means uncommon. We remember to have seen one in a creek in North Carolina, the water of the creek blocks of any required dimensions and shape beingquite salt.

should repeat them. If not then published, they may conclude that, for good reasons, the Editor declines them. The address of the writer should always be given.

Enquiries relating to patents, or to the patentablity of inventions, assignments, etc., will not be published here. All such questions, when initials only are given, are thrown into the waste basket, as it would fill half of our paper to print them all; but we generally take pleasure in answering briefly by mail, if the writer's address is given.

Hundreds of enquiries analogous to the following are sent : "Who sells dynamite, dualin, and lithofracteur? Who sells steam engine indicators? Why do not makers of watch-making machinery advertise in the SCIENTIFIC AMERICAN? Where can pure tallow, for cylinder lubrication, be obtained ?" All such personal inquiries are printed, as will be observed, in the column of "Business and Personal." which is specially set apart for that purpose, subject to the charge mentioned at the head of that solumn. Almost any desired infor-